

The Co-Sn-Zr (Cobalt-Tin-Zirconium) System

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Introduction

The Co-Sn-Zr system has been studied and one isothermal section has been established at reasonably low temperature and is reported here.

Binary Systems

The Co-Sn binary system [1983Nis, Massalski2] (Fig. 1) has three intermediate phases: Co_3Sn_2 in two allotropic forms Co_3Sn_2 (η) above $\sim 500^\circ\text{C}$ and Co_3Sn_2 (η_1) below 500°C , CoSn (π) and CoSn_2 (ζ). The η phase melts congruently at $\sim 1170^\circ\text{C}$ and the other two phases form through peritectic reactions: $\text{L} + \eta \leftrightarrow \pi$ at 936°C and $\text{L} + \pi \leftrightarrow \zeta$ at 525°C . Two eutectic reactions $\text{L} \leftrightarrow \gamma + \eta$ and $\text{L} \leftrightarrow \zeta + \beta$ occur at 1112°C and 229°C , respectively. The γ and β phases are the terminal solid solutions of fcc αCo and of tetragonal βSn , respectively. The $\gamma \leftrightarrow \varepsilon$ transformation possibly occurs through a eutectoid type reaction at $\sim 420^\circ\text{C}$ where ε is the terminal solid solution of close packed hexagonal (cph) εCo .

The Co-Zr system [Massalski2] (Fig. 2) shows the presence of five intermediate phases $\text{Co}_{11}\text{Zr}_2$ (ψ), Co_4Zr (ι), Co_2Zr (λ_2), CoZr (β_1), and CoZr_2 (ξ). The

probable presence of another phase CoZr_3 (ν) was reported but requires further confirmation. With the Co_2Zr phase being an exception, all other phases are stoichiometric compounds. The Co_2Zr , CoZr , and CoZr_2 phase are believed to melt congruently at approximately 1620 , 1420 , and 1125°C , respectively. The ψ and ι phases form through peritectic reactions: $\text{L} + \lambda_2 \leftrightarrow \psi$ at 1452°C and $\text{L} + \iota \leftrightarrow \psi$ at 1272°C . Four eutectic reactions occur in the Co-Zr system: $\text{L} \leftrightarrow \gamma + \psi$ at 1222°C , $\text{L} \leftrightarrow \lambda_2 + \beta_1$ at 1312°C , $\text{L} \leftrightarrow \beta_1 + \xi$ at 1061°C , and $\text{L} \leftrightarrow \xi + \alpha$ at 981°C where α is the body centered cubic (bcc) terminal solid solution of βZr . A eutectoid reaction $\alpha \leftrightarrow \xi + \varepsilon_2$ occurs at 834°C where ε_2 is the cph terminal solid solution of αZr . In the Co-rich region possibly a eutectoid reaction $\gamma \leftrightarrow \psi + \varepsilon_1$ occurs at $\sim 422^\circ\text{C}$.

The Sn-Zr system [Massalski2] (Fig. 3) has three intermediate phases, SnZr_4 (θ), Sn_3Zr_5 (ω), and Sn_2Zr (ϕ) of which the ω phase melts congruently at 1988°C . The probable existence of another phase Sn_4Zr_5 (T) has been suggested in the SnZr system. The θ and ϕ phases form through peritectoid and peritectic reactions: $\omega + \alpha \leftrightarrow \theta$ at 1327°C and $\text{L} + \omega \leftrightarrow \phi$ at 1142°C .

The reaction through which the T phase forms is not known. Two eutectic reactions $\text{L} \leftrightarrow \alpha + \omega$ and $\text{L} \leftrightarrow \phi + \beta$ occur at 1592 and $\sim 232^\circ\text{C}$, respectively. The cph ε_2 phase forms through a peritectoid reaction $\alpha + \theta \leftrightarrow \varepsilon_2$ at 982°C where ε_2 is the cph terminal solid solution αZr .

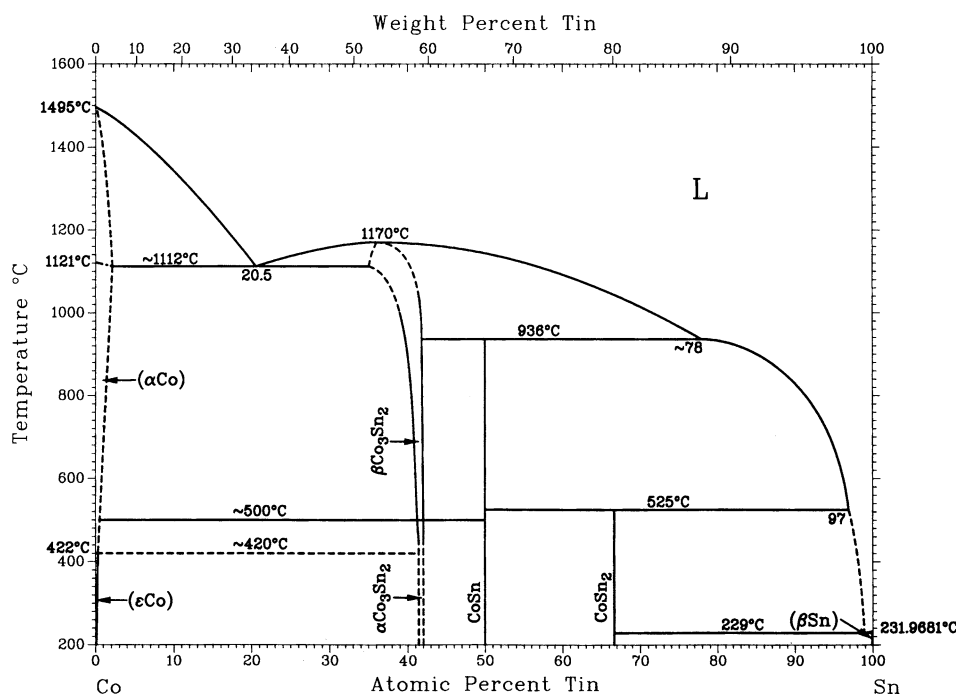


Fig. 1 The Co-Sn binary system [Massalski2]

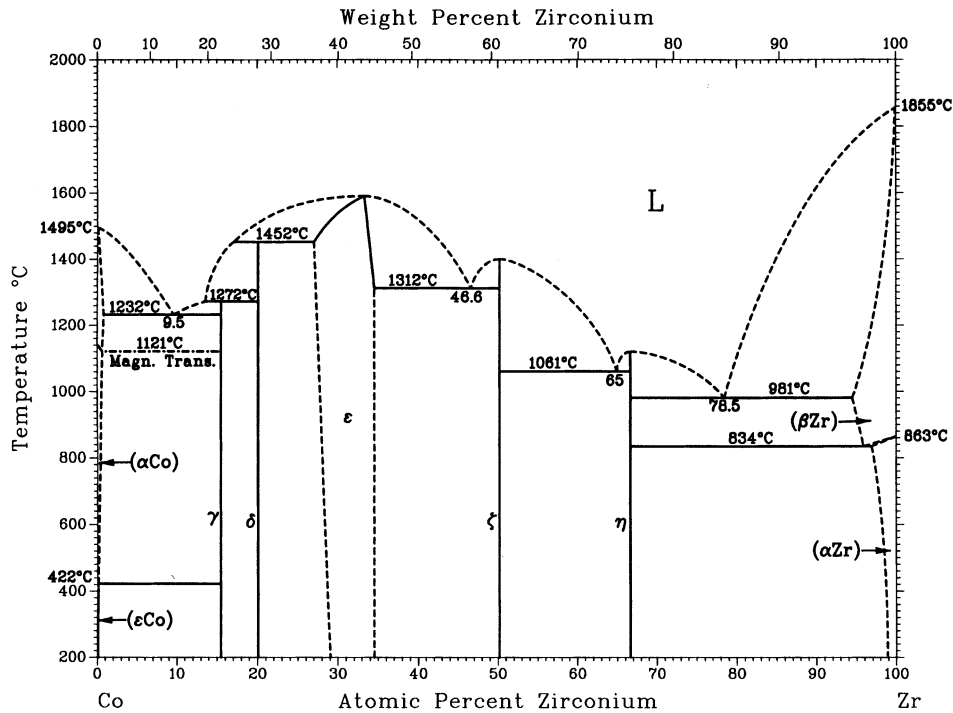


Fig. 2 The Co-Zr binary system [Massalski2]

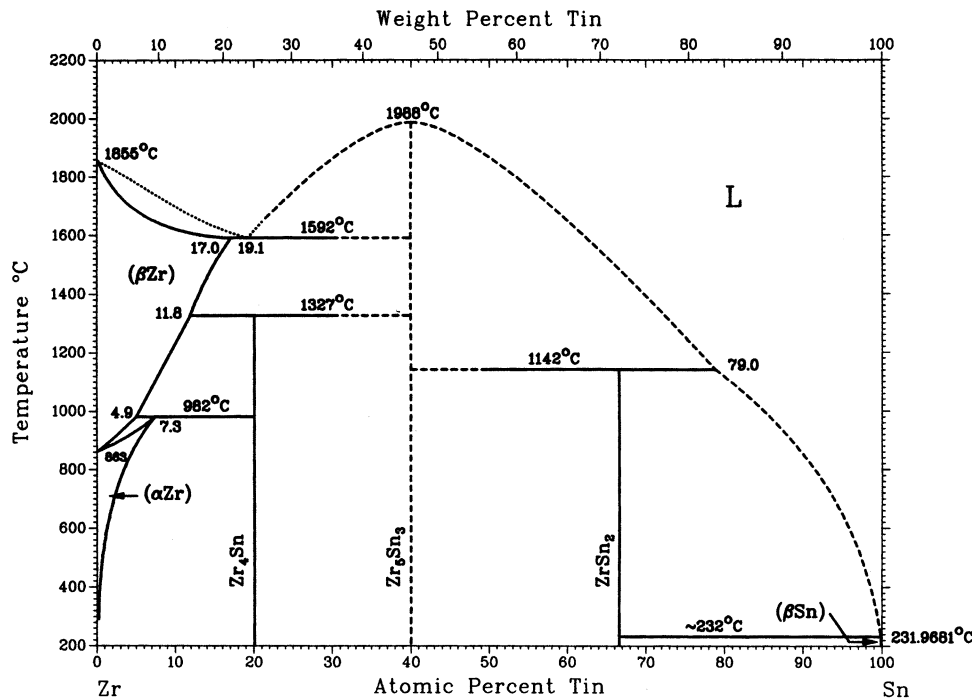


Fig. 3 The Sn-Zr binary system [Massalski2]

Binary and Ternary Phases

In the three binary systems Co-Sn, Co-Zr, and Sn-Zr given by [Massalski2] 11 intermediate phases form. Two

more phases, one is the Co-Zr system and the other is the Sn-Zr system, have been reported but possibly require further confirmation. In the Co-Sn-Zr system four ternary intermediate phases form. The binary and ternary phases of

Section II: Phase Diagram Evaluations

Table 1 Binary and ternary phases of the Co-Sn-Zr system and their structure data

| Phase designation | Composition | Pearson's symbol | Space group | Type | Lattice parameters, nm | | |
|-------------------|-----------------------------------|------------------|-----------------------------------|---------------------|------------------------|----------|----------|
| | | | | | <i>a</i> | <i>b</i> | <i>c</i> |
| γ | (α Co) | <i>cF4</i> | <i>Fm$\bar{3}m$</i> | Cu | ... | ... | ... |
| ε_1 | (ε Co) | <i>hP2</i> | <i>P6$_3$/mmc</i> | Mg | ... | ... | ... |
| α | (β Zr) | <i>CI2</i> | <i>Im$\bar{3}m$</i> | W | ... | ... | ... |
| ε_2 | (α Zr) | <i>hP2</i> | <i>P6$_3$/mmc</i> | Mg | ... | ... | ... |
| β | (β Sn) | <i>tI4</i> | <i>I4$_1$/amd</i> | β Sn | ... | ... | ... |
| η | Co $_3$ Sn $_2$ (HT) | <i>hP4</i> | <i>P6$_3$/mmc</i> | AsNi | 0.411 | ... | 0.5183 |
| η_1 | Co $_3$ Sn $_2$ (LT) | <i>oP20</i> | <i>Pnma</i> | Ni $_3$ Sn $_2$ | ... | ... | ... |
| π | CoSn | <i>hP6</i> | <i>P6/mmc</i> | CoSn | 0.5279 | ... | 0.4259 |
| ζ | CoSn $_2$ | <i>tI112</i> | <i>I4/m</i> | Al $_2$ Cu | 0.6361 | ... | 0.5452 |
| ψ | Co $_{11}$ Zr $_2$ | ... | ... | ... | ... | ... | ... |
| ι | Co $_4$ Zr | <i>cF116</i> | <i>Fm$\bar{3}m$</i> | Mn $_{23}$ Th $_6$ | 1.1516 | ... | ... |
| λ_2 | Co $_2$ Zr | <i>cF24</i> | <i>Fd$\bar{3}m$</i> | Cu $_2$ Mg | 0.69512 | ... | ... |
| β_1 | CoZr | <i>cP2</i> | <i>Pm$\bar{3}m$</i> | CsCl | 0.3197 | ... | ... |
| ξ | CoZr $_2$ | <i>tI12</i> | <i>I4/mcm</i> | Al $_2$ Cu | 0.6364 | ... | 0.5518 |
| ν | CoZr $_3$ (a) | <i>oC16 hP8</i> | <i>Cmcm P6$_3$/mmc</i> | Re $_2$ BNi $_3$ Sn | ... | ... | ... |
| ϕ | SnZr $_2$ | <i>oF24</i> | <i>Fddd</i> | TiSi $_2$ | 0.957 | 0.564 | 0.992 |
| T | Sn $_4$ Zr $_5$ (a) | <i>hP18</i> | ... | GaTi $_5$ | ... | ... | ... |
| ω | Sn $_3$ Zr $_5$ | <i>hP16</i> | <i>P6$_3$/mcm</i> | Mn $_5$ Si $_3$ | 0.8461 | ... | 0.5795 |
| θ | SnZr $_4$ | <i>cP8</i> | <i>Pm$\bar{3}m$</i> | Cr $_3$ Si | 0.565 | ... | ... |
| ψ | CoSnZr | ... | <i>P6$_2$m</i> | AlNiZr | 0.7133 | ... | 0.3571 |
| Ω | Co $_2$ SnZr | ... | <i>Fm$\bar{3}m$</i> | AlCu $_2$ Mn | 0.6227 | ... | ... |
| Δ | Co $_{1.65}$ Sn $_{1.35}$ Zr $_6$ | ... | <i>P6$_2$m</i> | ... | 0.7971 | ... | 0.3453 |
| Φ | Co $_6$ Sn $_{18}$ Zr $_5$ | ... | ... | ... | 1.3268 | ... | ... |

(a) Probably exists

the Co-Sn-Zr system and their structure data are given in Table 1.

Ternary System

The Co-Sn-Zr system has been studied by [1995Sta] using 125 alloys, arc melted under purified argon atmosphere. High purity component elements, electrolytic Co of 99.6 mass% purity, Sn of 99.99 mass% purity, and iodide Zr of 99.9 mass% purity, were used to prepare the alloys. For annealing, the alloys were sealed in evacuated quartz capsules. The alloys containing ≤ 40 at.% Sn were first annealed at 797 °C for 240 h and subsequently annealed at 497 °C for 240 h. The alloys with > 40 at.% Sn were directly annealed at 497 °C for 240 h. The alloys after annealing were quenched in cold water. For phase analysis and phase identification, a Debye Scherrer x-ray powder diffraction camera was used. For a few ternary intermediate phases structure analysis was done by using Laue and rotating crystal techniques and for structure refinement a diffractometer was used.

The 497 °C isothermal section of Co-Sn-Zr system established by [1995Sta] is given in Fig. 4. Four ternary intermediate phases were found of which two of the phases Co-Sn-Zr (ψ) and CoSnZr (Ω) were reported earlier by

[1986Sko] and [1976Sob], respectively. The two new ternary intermediate phases were found to exist with approximate compositions of CoSn $_3$ Zr (ϕ) and CoSnZr $_4$ (Δ). The binary intermediate phases were found to have very limited solubility ~ 1 at.% of a third element. Even though the binary diagrams of Co-Zr and Sn-Zr systems do not include the CoZr $_3$ (ν) and Sn $_4$ Zr $_5$ (γ) phases, the 497 °C isothermal section by [1995Sta] show these two phases. These two phases were found in equilibrium with the ternary intermediate phases Δ and ψ , respectively. The CoSnZr (Ω) phase region is a small elongated region extending along a line joining Co and Sn $_{50}$ Zr $_{50}$ composition, extending from $\sim 40\%$ ~ 50 at.% Co. The Ω phase was found in equilibrium with the λ_2 , η_1 , π , ζ , ϕ , ψ , and ω phases. The ψ phase was found in equilibrium with the Φ , ϕ , T, Ω , and ω phases. The Φ phase was reported to be in equilibrium with the Ω , ψ , ζ , Sn, and ϕ phases. Since at 497 °C Sn is in liquid form the Φ phase should be in equilibrium with liquid Sn. The Δ phase was found in equilibrium with the β_1 , ξ , ν , θ , and ω phases. The Co $_{11}$ Zr $_2$ (ψ) phases exists in the Co-Zr binary system but the investigation by [1995Sta] does not show this phase in the isothermal section at 497 °C. The Co-corner of the Co-Sn-Zr thus appears to be incomplete and should be reinvestigated. The binaries Co-Zr system and Sn-Zr system also should be carefully studied to establish the existence as well as the mode of formation of the CoZr $_3$ and Sn $_4$ Zr $_5$ phases.

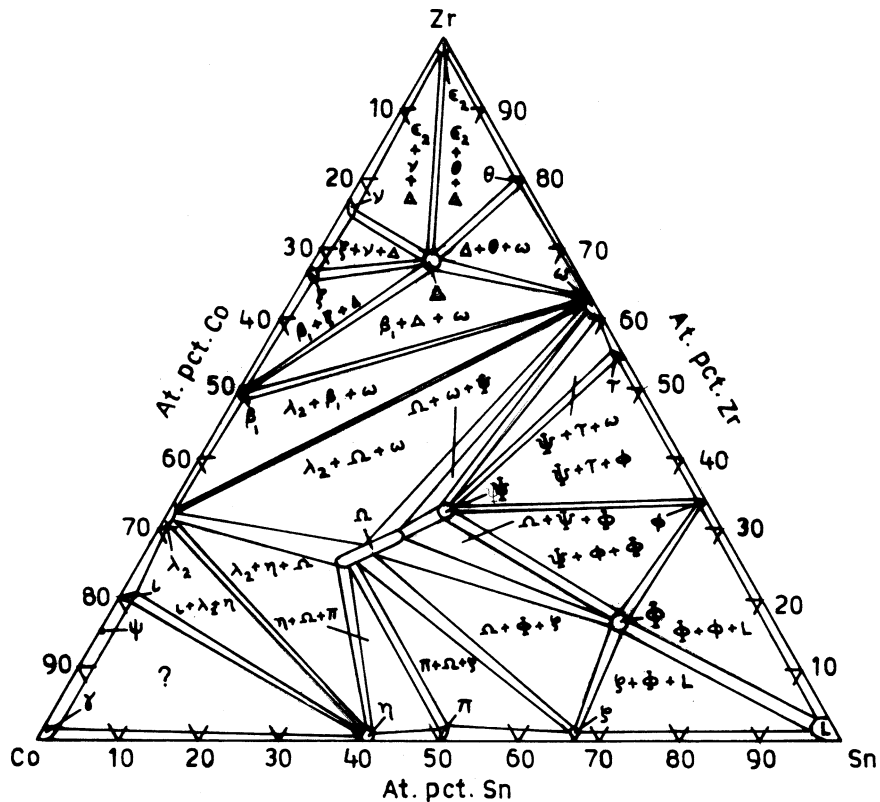


Fig. 4 Isothermal section of the Co-Sn-Zr system at 497 °C [1995Sta]

Of the four ternary intermediate phases found in the Co-Sn-Zr system, two intermediate phases CoSnZr and Co_2SnZr were reported earlier by [1974Dwi] and [1976Sob], respectively. The CoSnZr phase was reported to be of Fe_2P type with lattice parameters $a = 0.7156$ nm and $c = 0.3563$ nm and the Co_2SnZr phase was reported to be of BiF_3 type with lattice parameter $a = 0.6285$ nm. [1986Sko], however, reported the CoSnZr phase to be of AlNiZr type, a superstructure of Fe_2P type structure, with lattice parameters $a = 0.7133$ nm, $c = 0.3571$ nm, and $\gamma = 120^\circ$. [1986Sko] also showed that the Co_2SnZr phase to be of AlCu_2Mn type with lattice parameter $a = 0.6234$ nm. Laue and rotation x-ray diffraction patterns taken with a single crystal extracted from the CoSnZr_4 alloy showed that this phase is of Fe_2P type with lattice parameter $a = 0.7971$ nm and $c = 0.3453$ nm. Structure refinement was done using a diffractometer. The structure was found to be a superstructure of Fe_2P type structure and the ideal composition was reported to be $\text{Co}_{1.65}\text{Sn}_{1.35}\text{Zr}_6$. An x-ray diffraction pattern of CoSn_3Zr phase was found to closely resemble the diffraction pattern of a phase found in the Er-Rh-Sn system having a composition $\text{Sn}_{1-x}(\text{Sn}_{1-x}\text{Zr}_x)\text{Zr}_4\text{Rh}_6\text{Sn}_{18}$ which has a cubic structure. The diffraction pattern of an alloy with composition $\text{Co}_{20}\text{Sn}_{65}\text{Zr}_{15}$ could be indexed reasonably well with a cubic cell with lattice

parameter $a = 1.3682$ nm. On the basis of similarity of this diffraction pattern of the $\text{Co}_{20}\text{Sn}_{65}\text{Zr}_{15}$ alloy with that of the Er-Rh-Sn phase, the composition for the CoSn_2Zr phase has been given a tentative composition identification of $\text{Co}_6\text{Sn}_{18}\text{Zr}_5$ phase. Further work has to be done to identify the $\text{Co}_6\text{Sn}_{18}\text{Zr}_5$ phase.

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